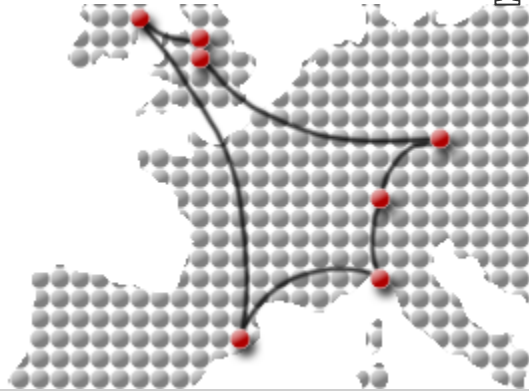




RESEARCH & INNOVATION
Marie Curie Actions

Kick off meeting

MagNETic FUN



Barcelona
23-24 February 2013

Nanotoxicology

Robert N. Grass

ETH Zurich, Switzerland

TurboBeads GmbH, Zurich, Switzerland

The problem with nanotoxicology - generalization

Penetration of Intact Skin by Quantum Dots with Diverse Physicochemical Properties

Jessica P. Ryman-Rasmussen, Jim E. Riviere, and Nancy A. Monteiro-Riviere¹

Center for Chemical Toxicology Research and Pharmacokinetics, College of Veterinary Medicine, North Carolina State University, Raleigh, North Carolina 27606

Test of 1 spherical QD (4.6 nm) and 1 ellipsoid QD (6x12 nm) with 3 surface coatings

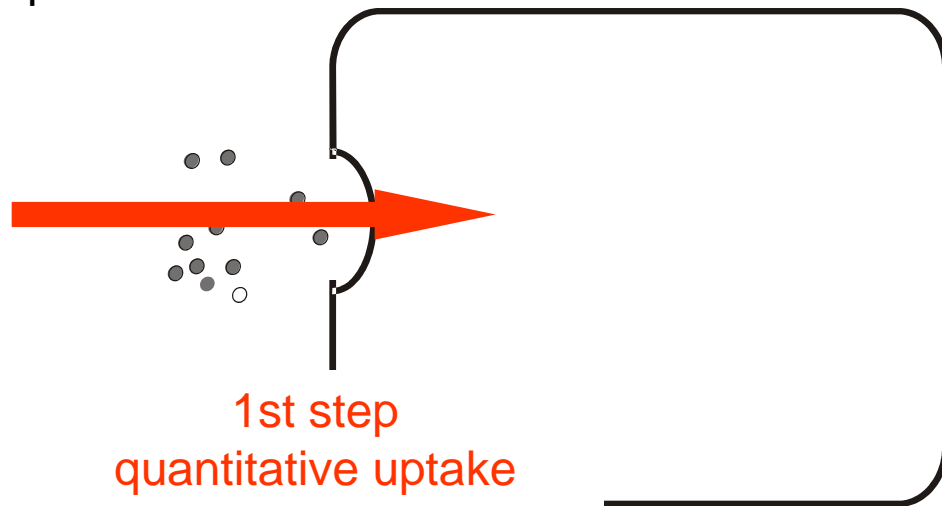
The problem with nanotoxicology

“The penetration of QD through skin was addressed by the Monteiro-Riviere research group employing an ex vivo porcine skin model. Their **initial results found that**, with 8 and 24 h of exposure to QD, porcine skin exhibited **QD penetration throughout the epidermis** and deep into the dermis in some cases. In a more **recent follow-up study, they** reported **contrasting results** in that minimal penetration of QD through ex vivo porcine skin was found, with the bulk of the QD remaining in the stratum corneum. Reasons for this discrepancy remain unclear, however, **other researchers** have examined the question of skin penetration employing different NP types (metals, polymers) using ex vivo skin models, **and again contrasting** results of both high and low levels of NP penetration are reported.”

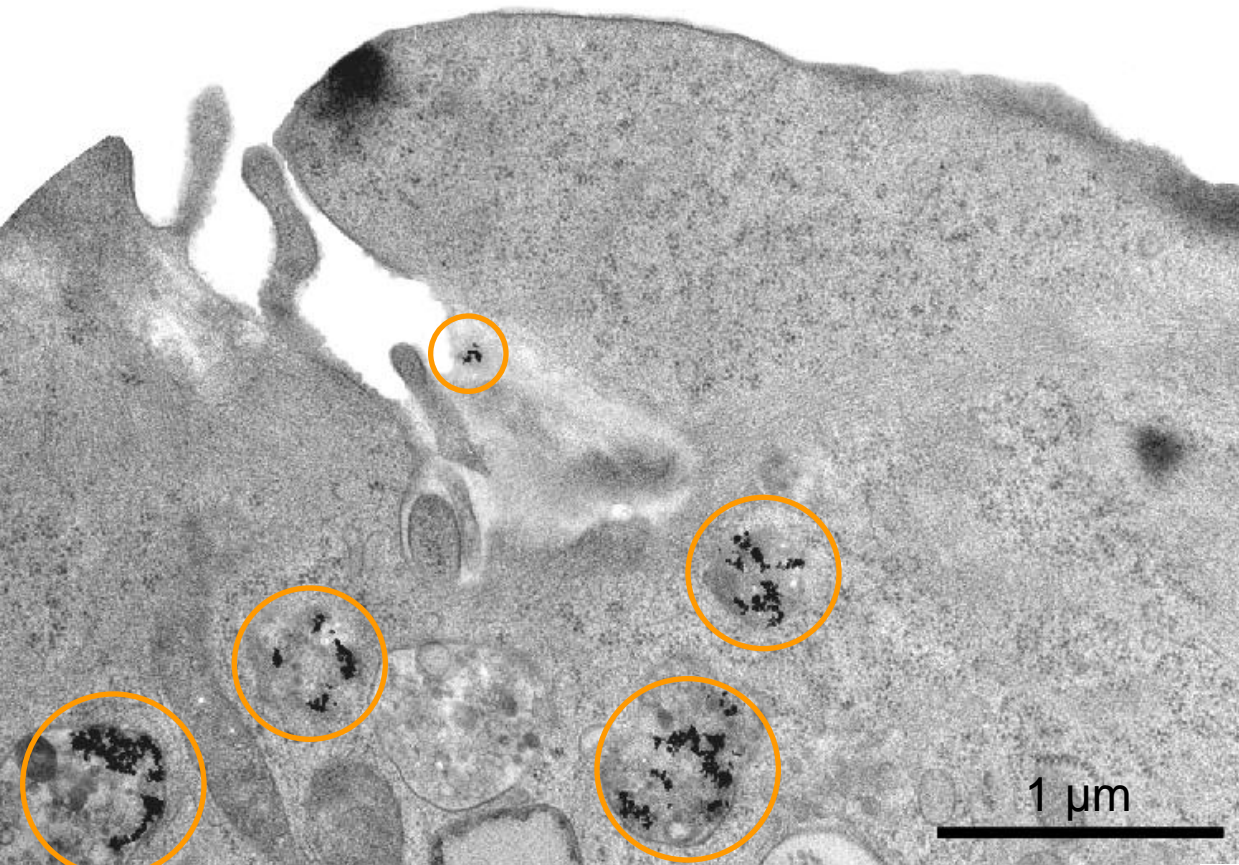
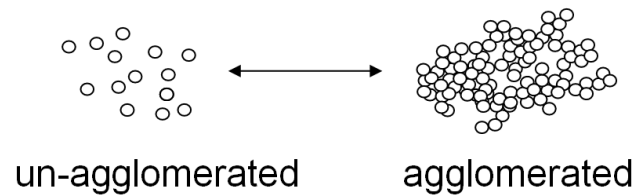
NANO LETTERS
2008 Vol. 8, No. 9 2779-2787

Why we are afraid: Nanoparticle in living cells?

uptake of nanoparticles



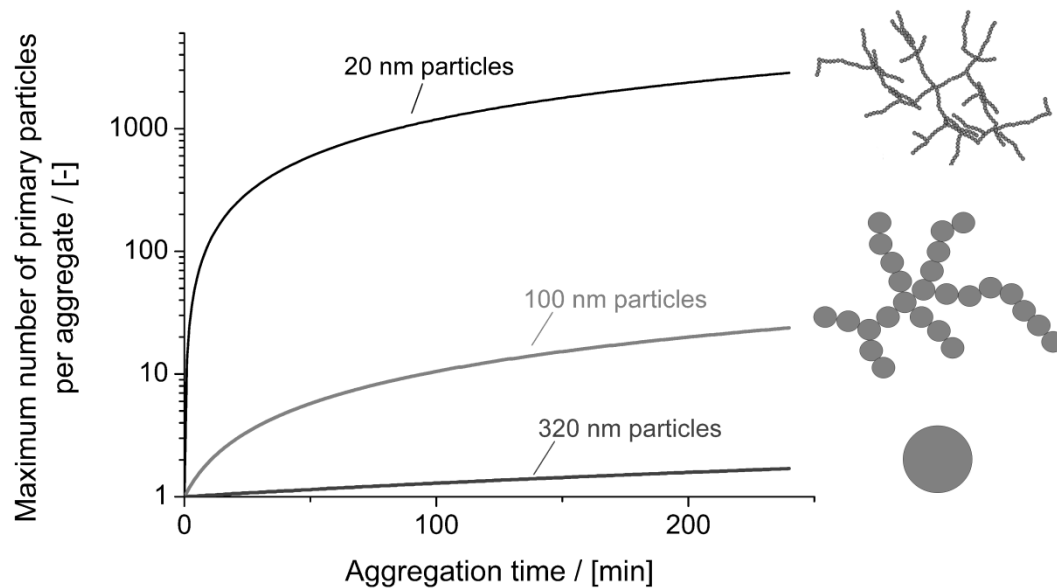
quantitative uptake of nanoparticles in cells



$$\frac{dn_t}{dt} = -\frac{1}{2} \frac{\beta}{W} n_t^2$$

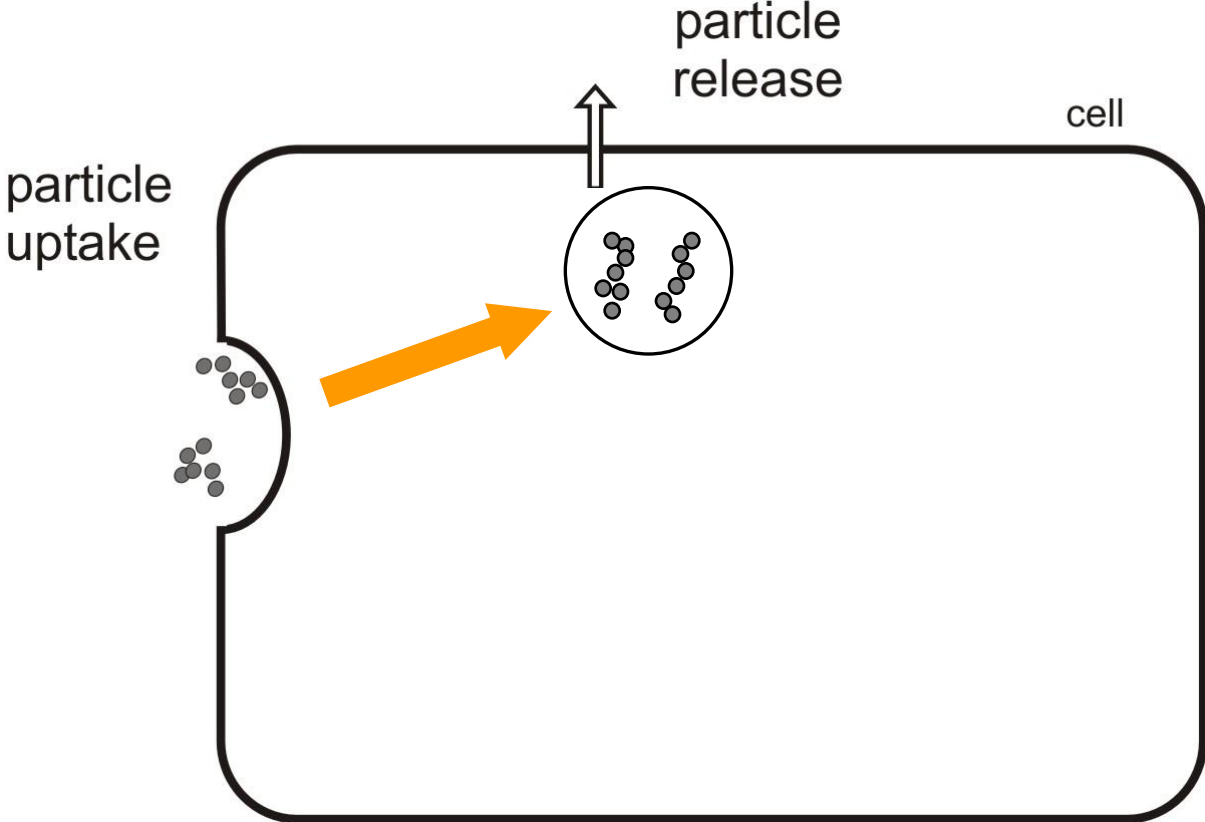
Friedlander (1953), Smoke, Dust and Haze

agglomeration – same mass concentration different size

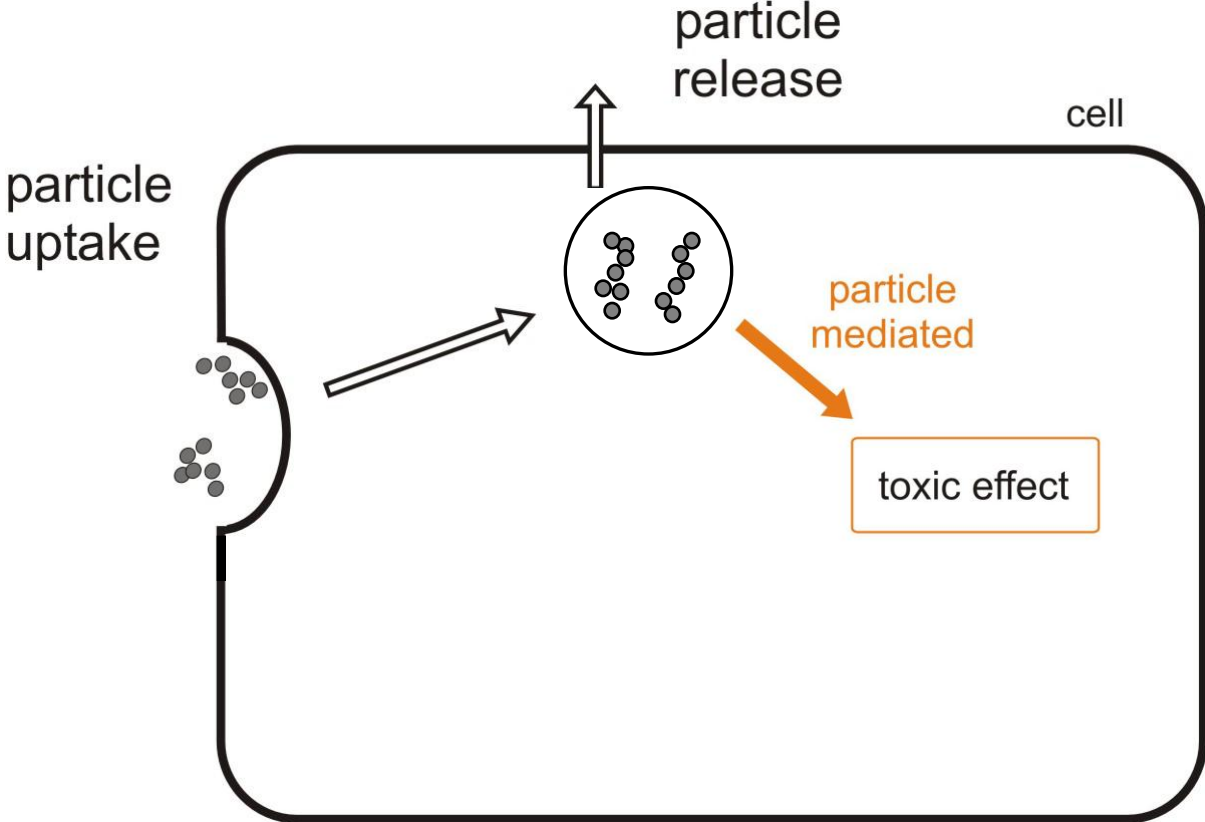


Biodistribution = f (time, concentration)

Schematic diagram



Schematic diagram



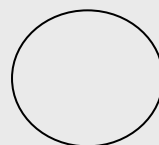
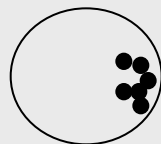
Trojan horse mechanism



ZnO-nanoparticle



pH = 5.5



Zn²⁺-ions



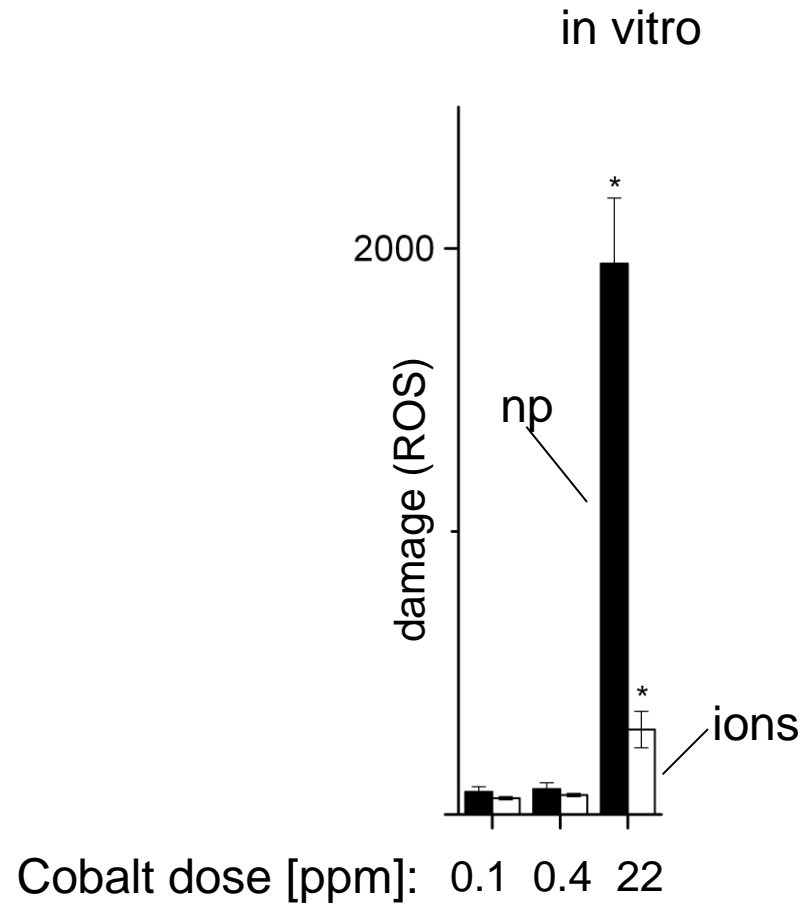
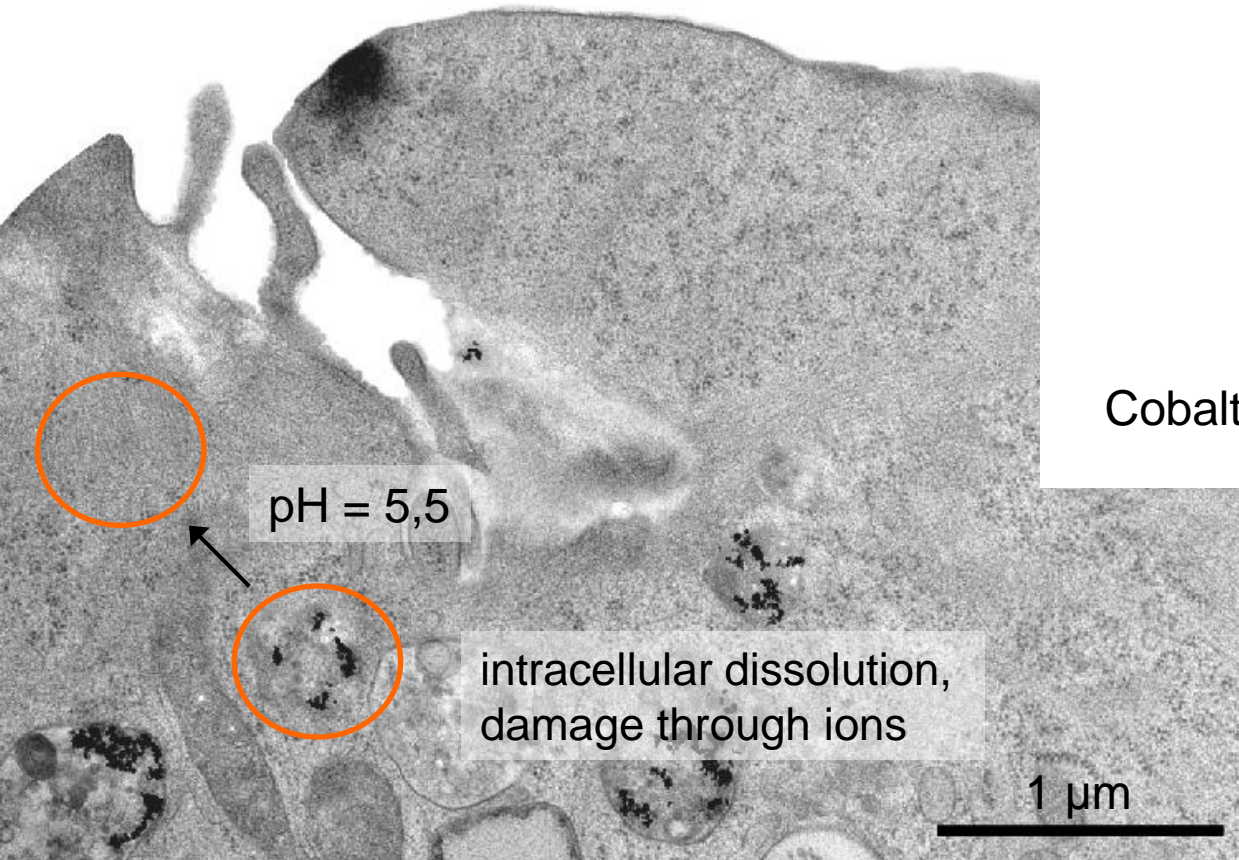
ZnO-microparticle



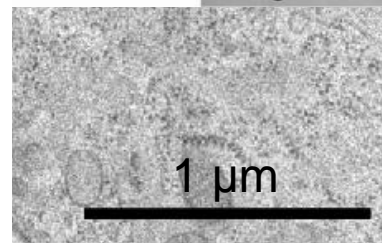
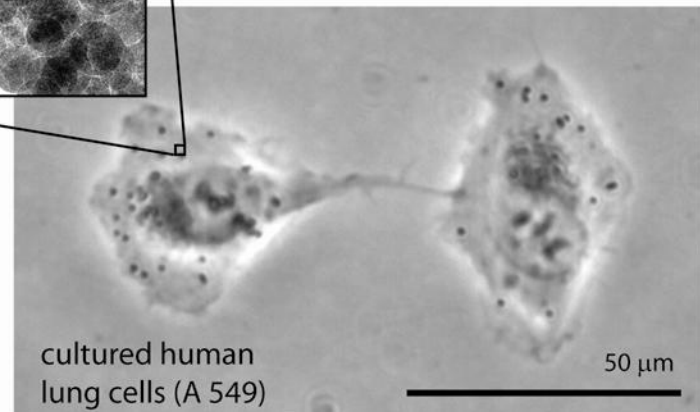
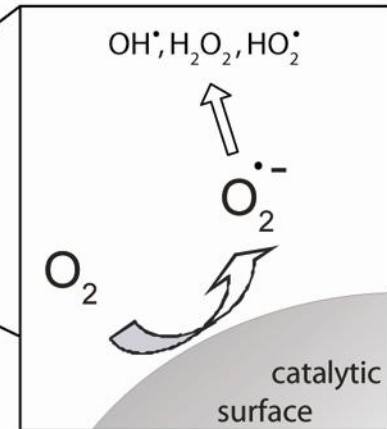
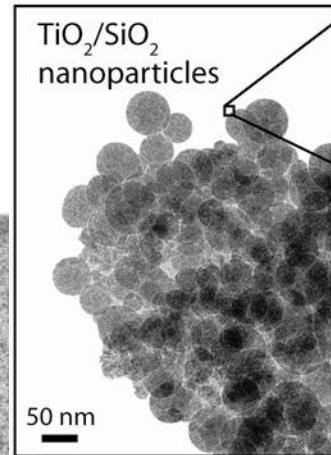
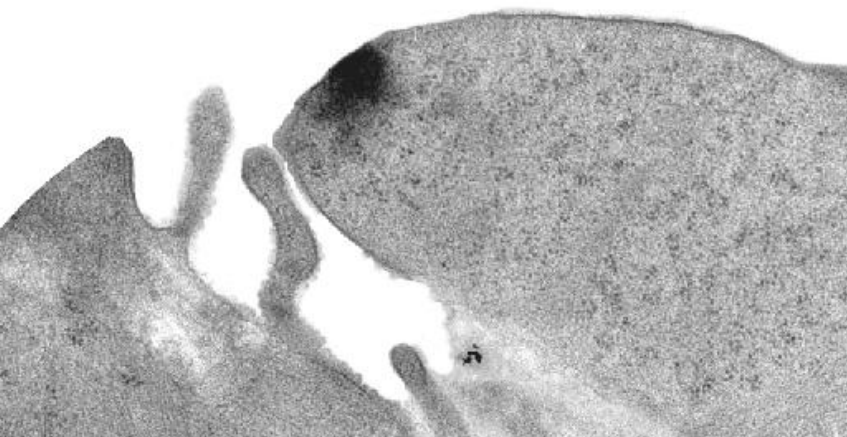
cell

Intracellular dissolution
additional damage through ions

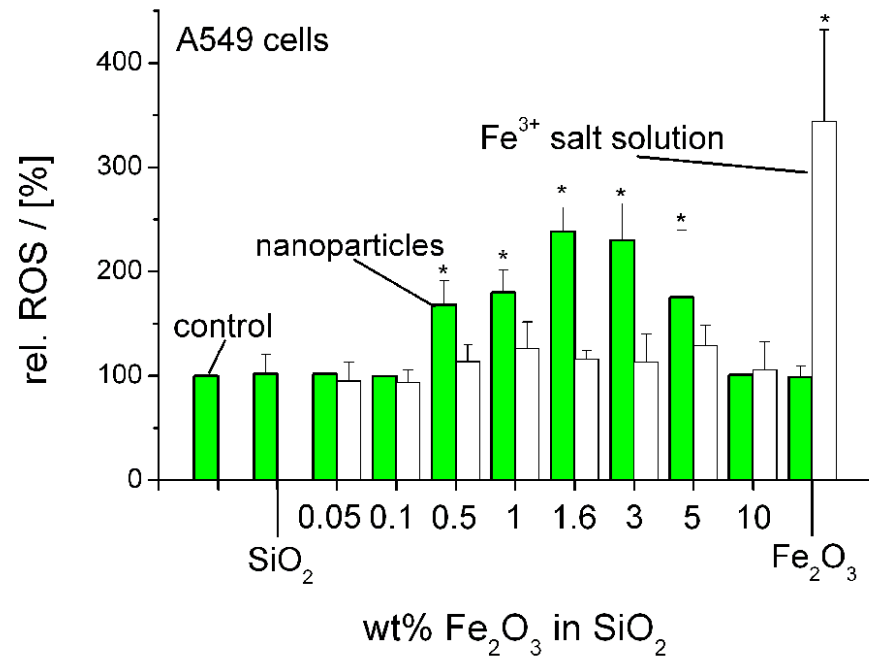
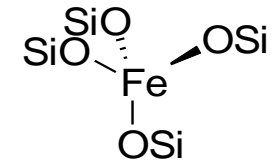
trojan horse-type enhanced heavy metal uptake



heterogeneous Catalysis inside living cells



heterogeneous catalysis inside living cells



Classical toxin

- Bio-distribution
- Action
- Clearance
- Dose/Effect relations

Nanoparticles

- Agglomeration, diffusion and sedimentation change distribution kinetics
- Action; Solubility alters release
- Clearance
- Dose/effect
- Catalytic effects: Non-mass constant toxic action

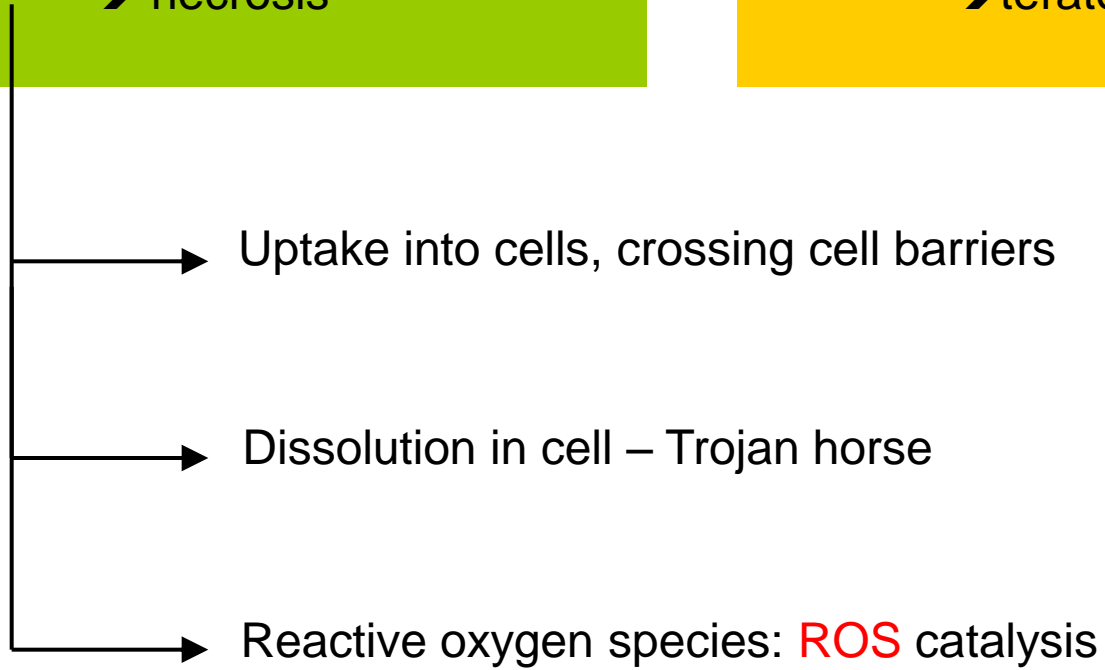
Nanoparticle associated risk

acute effects

- apoptosis
- necrosis

long term effects

- mutagenicity
- teratogenicity



Nano is not new

Global commodities:

- Silica np. (food additive)
 - Titania np. (white paint, toothpaste, sunscreen)
 - Carbon black np. (car tires, printing inks)
- Safely used today in consumer goods

Fullerenes, carbon nanotubes (asbestos)

known for toxicity

Risk of nano products

risk := damage potential * occurrence probability

linking to measurable values

Material properties

- solubility
- degradation
- fate
- activity
- behavior

...

acute or long term effects

occurrence probability

- mass, sources
- exposure concentration
- frequency
- transport
- target tissue

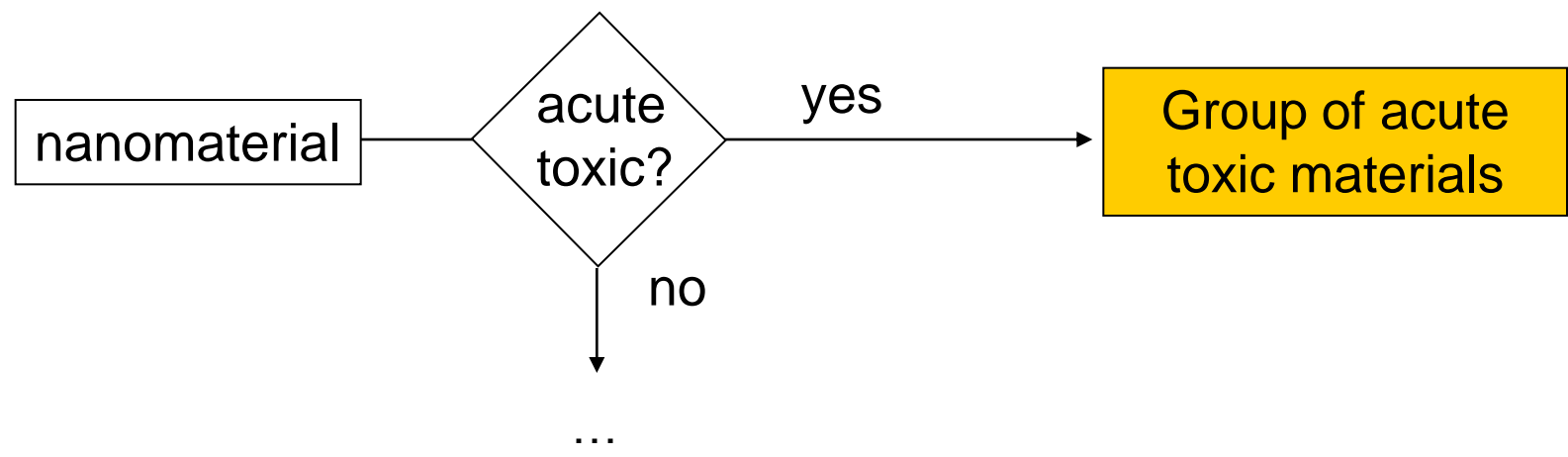
...

exposure and uptake

risk := damage potential * occurrence probability

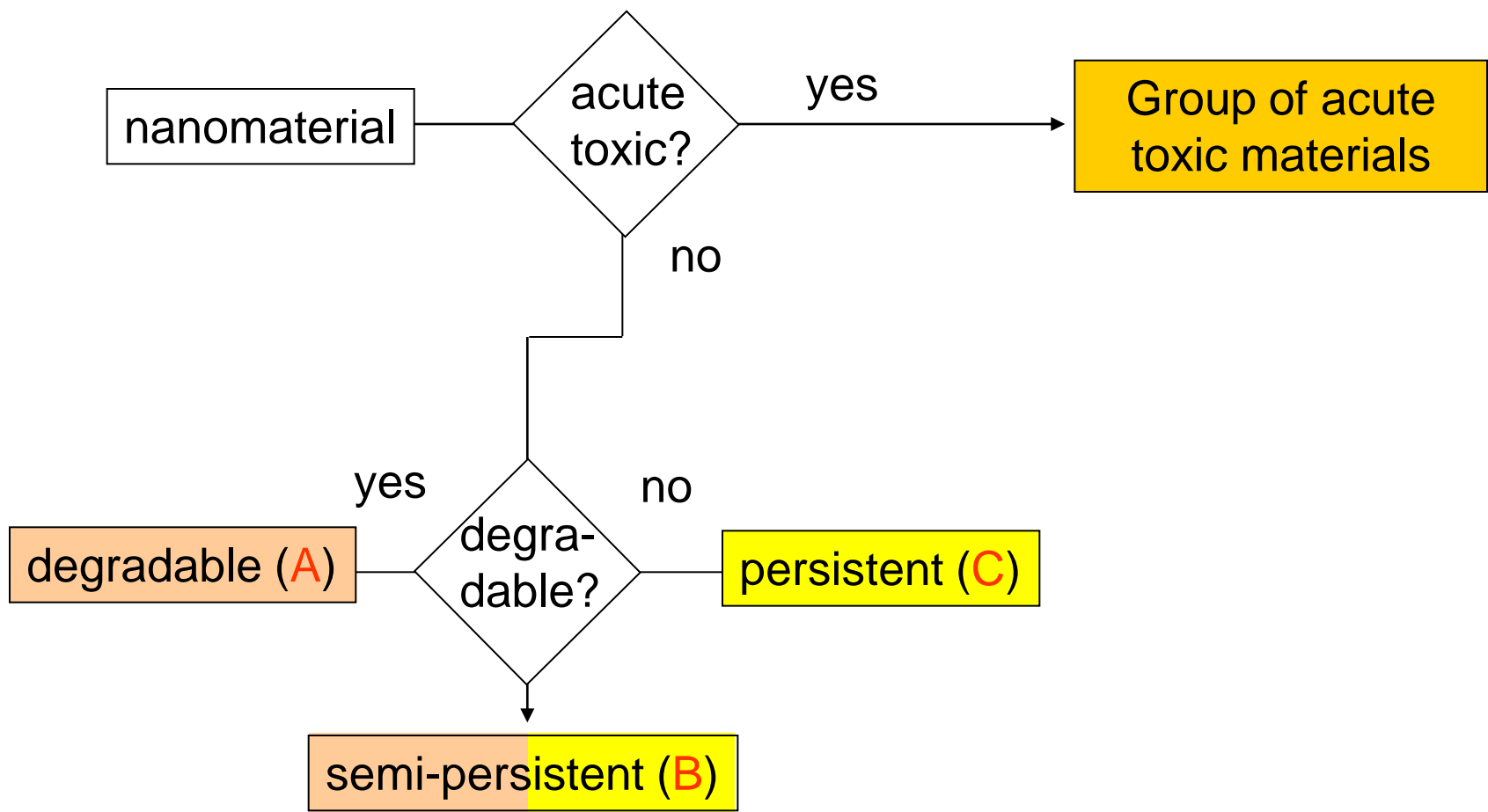
1st criterion -

is it acute toxic?



2nd criterion -

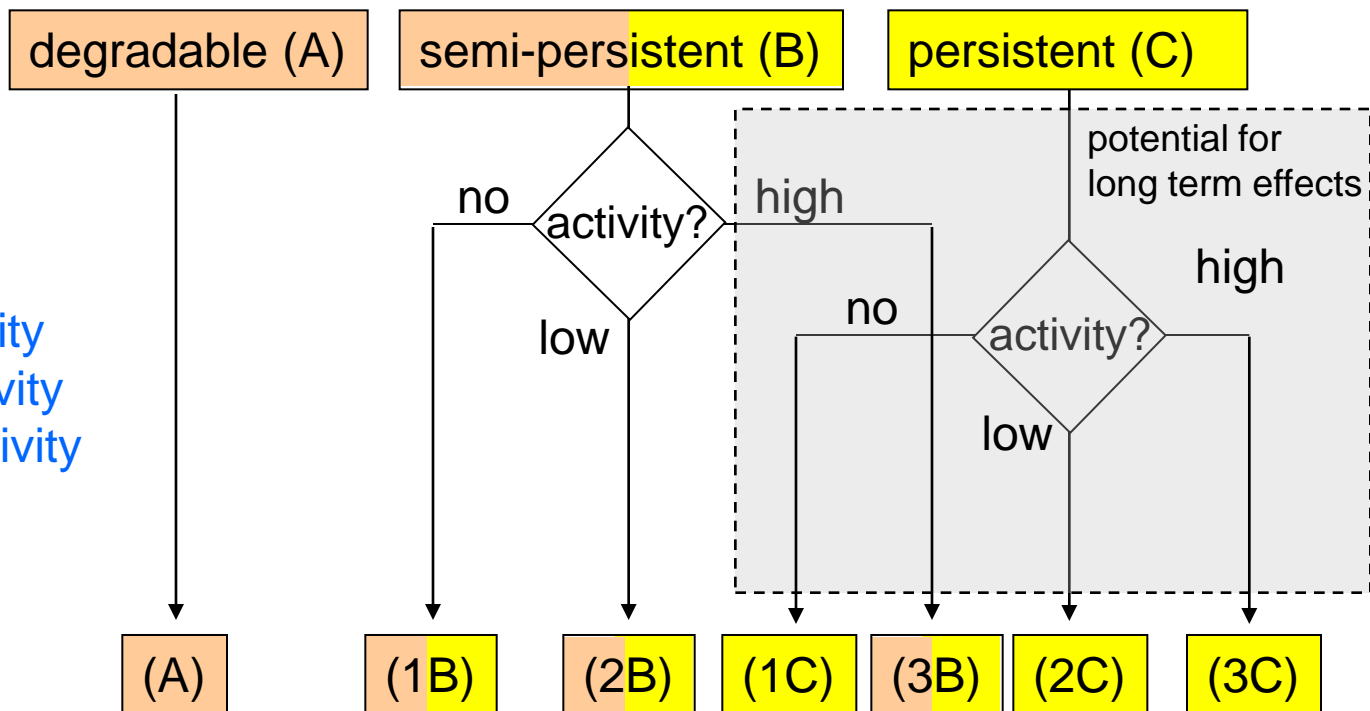
is it degradable?



3rd criterion -

chemical activity

- 1 - no chemical activity
- 2 - low chemical activity
- 3 - high chemical activity




Biological damage



Damage potential

 ZnO **high** ⇒ acute toxic

 Ca₃PO₄ **low** ⇒ low damage potential

 CNT **high** ⇒ potential long term effects

OVERALL RISK depends on the use (**occurrence probability**)

Occurrence probability

risk := damage potential * occurrence probability

Who is exposed?

With what ?

- an aerosol
- a liquid dispersion
- a dry powder
- a solid nanocomposite (e.g. np in polymer)

Exposure time, route of entrance

Think about

- Nanoparticles as food additive
- Nanoparticles in cosmetics
- Nanoparticles in aerosol spray can
- Nanoparticles embedded into battery material
- Nanoparticles in technical catalyst
- Nanoparticles as cancer treatment



In the laboratory routine

- Nanoparticles dispersed in **liquids**
 - **Treat as toxic** liquid chemicals
 - Wear gloves and safety glasses at all times
 - Wash up spills wearing gloves
- Nanoparticles don't translocate through healthy skin
- Use standard chemical/biochemical laboratory practice
- Don't wear gloves all the time
 - Doors, drawers, computers, labbook, pens only without gloves

In the laboratory routine

- **Dry** nanopowders (agglomerated)
 - Treat in well ventilated areas (i.e. **hoods**) **only**
 - Weighing, grinding, pouring etc.
 - Nanoparticles don't deagglomerate
 - Once stuck together as a powder, they stay together
 - Unless energy is introduced
 - Metals (even coated) are potentially pyrophoric
- Nanoparticle **aerosols**
 - Nanoparticle aerosols are taken up via the lung
 - In closed environments (**glove box** or similar) **only!**

Nanoparticle disposal

- Collect separately (including all solutions contacting nps)
- In dispersed (liquid) form
- Separated by
 - Aqueous
 - Non-aqueous
- Dispose as special waste via local authorities

Planning science with nanotoxicology

Before starting.....

- Think about product exposure
 - Manufacturing
 - Use of product/material
 - End of life
- A fast *in vitro* cell-test won't save you
 - Throwing nps on cells and measuring dead/alive is not scientific
 - Exposure scenario with correct cell lines, exposure pathway, biological endpoint, degree of aggregation and adapted concentrations **only**

Don't be afraid...

.... be cautious!

Don't generalize

.... differentiate

Further Reading: Stark WJ, Nanoparticles in Biological systems,
Angew. Chem. Int. Ed. 2011, **50**, 1242-58.